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# **CVD** Growth of Two-Dimensional *MoS*<sub>2</sub> Depending on the Location of the *SiO*<sub>2</sub> Substrates

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**Abstract.** Transition metal dichalcogenides, like  $MoS_2$ , were attracted attention due to their optoelectronic properties. But fabrications of such materials still a challenge task. In this work, chemical vapor deposition (CVD) was used to synthesize two-dimensional  $MoS_2$  from S and  $MoO_3$  powders on  $SiO_2$ . Lateral size of domains were up to 80 mkm and thickness of single-layered  $MoS_2$  was 0.9 nm. The structural and optical properties of the obtained two-dimensional  $MoS_2$  are studied. We find that the formation of  $MoS_2$  domains with different size is dependent on location of the  $SiO_2$  substrates.

#### **INTRODUCTION**

After the discovery of graphene, great progress has been made in obtaining other two-dimensional materials. Today, such materials, like the class of transition metal dichalcogenides, in particular molybdenum disulphide ( $MoS_2$ ) shows great interest [1,2]. Molybdenum disulphide is one of the most promising semiconductor materials for nanoelectronics, optics and spintronics due to its unusual electronic and optical properties [3]. Multilayer  $MoS_2$  is a semiconductor with an indirect 1.2 eV band gap. However, the most interesting properties of  $MoS_2$ 2 appear in the study of single-layer  $MoS_2$ , which is a 1.8 eV direct band gap semiconductor [4].

At present two-dimensional  $MoS_2$  is obtained by mechanical exfoliation, electrochemical intercalation of lithium ions, molecular beam epitaxy, atomic-layer deposition and other methods [5]. But the most promising method is chemical vapor deposition (CVD), which makes it possible to obtain uniform two-dimensional  $MoS_2$  films of large area, which is an urgent task for industrial production [6].

#### **METHODS**

For the growth of  $MoS_2$ , silicon with an oxide film of  $SiO_2$  300 nm in thickness was used as substrates. Before the growth process, the surfaces of the substrates were pretreated in an ultrasonic bath with acetone and isopropyl alcohol for 5 minutes to remove contaminants. To remove organic bonds from the surface, the substrates were treated in a sulfuric peroxide solution in a ratio of 1:3, followed by washing in deionized water. The substrates were then processed in an oxygen plasma for 5 minutes at a power of 150 W.

The synthesis of molybdenum disulphide was carried out by the CVD method in the three-zone ceramic furnace Nabertherm 80/750/RS, schematically depicted in Fig. 1. In a quartz tube 5 cm in diameter, crucibles with precursors and samples were placed at a certain distance from each other as shown in Fig.1 This tube was subsequently introduced into a large ceramic tube of the furnace. The starting precursors for the synthesis were  $MoO_3$  and sulfur powders of Sigma Aldrich in a weight ratio of 1:13. The substrates on which  $MoS_2$  was grown were located at different distances from the precursor  $MoO_3$ . To reduce the amount of evaporated  $MoO_3$ , the crucible was covered with a silicon substrate. For the controlled synthesis, parameters were chosen: temperature, time, pressure, argon supply rate, initial precursor concentrations and location of substrates relative to precursors. Synthesis process includes:

1. The pre-vacuum pump is evacuated to 10-5 bar for 1 hour.

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FIGURE 1. Schematic illustration of the CVD growth setup of SiO<sub>2</sub>

2. Inert gas (Ar) is fed into the chamber at a flow rate of  $100 \text{ } cm^3/min$  for 1 hour to a pressure of 1 atm in the chamber when the temperature reaches  $300^{\circ}$ C.

3. The chamber is heated to a synthesis temperature of 700°C. 4. The growth of  $MoS_2$  on  $SiO_2$  occurs in the Ar flow at a rate of 10  $cm^3/min$  for 30 minutes at 700°C.

5. After synthesis, the chamber was naturally cooled to room temperature in the Ar stream to  $300 \text{ cm}^3/\text{min}$ .

Optical images were taken with an optical microscope Nikon eclipse LV100. The morphology of the surface of  $MoS_2$  films grown by CVD on the  $SiO_2$  substrate was studied by scanning probe microscope AFM / STM "SolverNext" and scanning electron microscope JEOL JSM-7800F. Measurements of the Raman spectra of  $MoS_2$  were carried out at room temperature using NT-NTRA SPECTRA from NT-MDT. The spectra were excited by a semiconductor laser with an excitation wavelength of 532 nm. Elemental analysis of the composition of grown  $MoS_2$  films was carried out by Oxford Instruments Energy-Dispersive Spectroscopy (EDS).

#### **RESULT AND DISCUSSION**

Optical measurements showed the growth of domains of  $MoS_2$  on the  $SiO_2$  surface in the form of triangles (Fig. 2). Depending on the location of the substrates relative to the  $MoO_3$  precursor, domains with different lateral sizes were obtained. Samples were located 5-7 cm from the precursor  $MoO_3$ , domains were 10-20 mkm. With decreasing the distance between samples and precursor domain size was increased up to 40 mkm. The largest  $MoS_2$  triangles were obtained, when samples located directly above the precursor. According to our results, larger flakes were grown near Mo precursor because of higher concentration  $MoO_3$  and different gas flow through the surfaces of substrates.



**FIGURE 2.** Optical images of domains grown at different distances from the precursor  $MoO_3$ : (a) more than 5 cm, (b) more than 2 cm, (c) directly above the precursor

AFM measurement showed 0.9 nm thickness of one layer of molybdenum disulphide (Fig. 3). Also, the surface of the obtained  $MoS_2$  was studied, where the presence of defects in the center of some domains was shown, and overgrown fragments of the same material on the edges of the domains were found.

Figure 4 shows the Raman spectra of  $MoS_2$  with two most intense Raman peaks in the 380 and 400  $cm^{-1}$  region were observed. The frequency difference between these peaks allowed to determinate the number of layers of grown



FIGURE 3. AFM image of a single-layer  $MoS_2$  domains on the  $SiO_2$  surface and the height profile along the line S1

 $MoS_2$ . For single-layer  $MoS_2$ , the frequency difference between the peaks was  $\Delta v = 19 \ cm^{-1}$ , for the bi-layer  $\Delta v = 21 \ cm^{-1}$  and for the multi-layer  $\Delta v = 25 \ cm^{-1}$ .



FIGURE 4. Raman spectra of synthesized single-layer, two-layer and multilayer MoS<sub>2</sub>

The data of the energy dispersive X-ray spectral microanalysis, which is shown in Fig. 6, made it possible to determine the elemental composition of the samples-molybdenum (0.19 at.%), sulfur (0.48 at.%), silicon (28.41 at.%), oxygen (70, 91 at.%). It should be noted that the spectral peaks of molybdenum and sulfur are located at values of about 2.3 keV and have only minor deviations with respect to each other, which made it difficult to identify this elements. Nevertheless, it was found that the atomic sulfur content is twice the atomic molybdenum content.



FIGURE 5. The line-scan EDX spectrum of a selected region from the STEM image (marked by a pink color rectangle)

#### CONCLUSIONS

Single-layer  $MoS_2$  with a thickness of up to 0.9 nm and lateral domain sizes up to 80 mkm was obtained on a  $SiO_2$  substrate by proposed CVD method. The domains with the largest lateral dimensions were obtained for samples were located less than 5 cm from the  $MoO_3$  precursor. It was found that with increasing distance the size of  $MoS_2$  domains decreases. According to our results, larger flakes were grown near Mo precursor because of higher concentration  $MoO_3$  and different gas flow through the surfaces of substrates.

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